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Chemical modification of the surface of bioactive glasses: Effects on apatite-forming ability. The influence of surface functionalization on the bioactivity of bioactive glasses (BG) was studied. BG with different chemical compositions, using strontium-calcium (Sr-Ca) and magnesium-calcium (Mg-Ca) as the main constituents, was activated at 800 degrees C for 5 h in air. BG without surface treatment acted as reference material. Functional groups attached to the surface of BG materials with different chemical compositions were confirmed by X-ray photoelectron spectroscopy (XPS) and Fourier transform infrared spectroscopy (FTIR). MG-CaB glass treated with H₂O₂ and reduced with Na₂S₂O₃ was modified with hydroxycarboxylic acid by a photochemical reduction process, and the surface chemistry was characterized by FTIR. Crystal structure of MG-CaB glass was not affected by the surface treatment. Formation of an apatite layer on the surfaces of BG was investigated in SBF by immersion for 4 weeks. The amount of apatite formed on the surface was determined by the atomic ratio of Ca/P. The in vitro apatite formation was affected by the chemical composition of the BG surface. Formation of the apatite layer was observed on the surfaces of Mg-CaB glasses, but not on the surfaces of Sr-CaBG. Reduction of the glass surface with Na₂S₂O₃ enhanced apatite formation. Apatite formed on the surface of Mg-CaB glass activated at 1200 degrees C for 15 min and reduced with Na₂S₂O₃ possessed hydroxyapatite (HA) crystal with a Ca/P ratio of 1.67.

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